

**DEPARTMENT OF PHYSICS
INDIAN INSTITUTE OF TECHNOLOGY, MADRAS**

PH3500 Classical Physics

Problem Set 8

26.10.2009

1. An atom with angular momentum quantum number j has $(2j + 1)$ equally-spaced energy levels in an applied magnetic field. If B denotes the magnitude of the field, these levels are given by $\varepsilon_r = g\mu_B Br$ where $r = -j, -j + 1, \dots, (j - 1), j$; g is a constant called the g -factor of the atom; μ_B is the Bohr magneton. Consider a paramagnetic material consisting of N such atoms, mutually non-interacting.

- (a) Show that the partition of the system is

$$Z = z^N \quad \text{where} \quad z = \left[\sinh(2j + 1)\xi / \sinh \xi \right], \quad \xi = \beta g \mu_B B / 2 .$$

- (b) Hence show that the magnetization $M = -\left(\partial F / \partial B\right)_T$ is given by

$$M = Ng\mu_B \left[\frac{1}{2}(2j + 1) \coth(2j + 1)\xi - \frac{1}{2} \coth \xi \right] .$$

The quantity in the square brackets is sometimes called the Brillouin function \mathcal{B}_j .

- (c) The isothermal paramagnetic susceptibility $\chi_T = \left[(\partial M / \partial B)_T \right]_{B=0}$ can then be calculated. Show that the answer is

$$\chi_T = Ng^2 \mu_B^2 j(j + 1) / (3k_B T) .$$

The effective number of Bohr magnetons of each atom (or ion) of such a paramagnetic substance is thus $g[j(j + 1)]^{1/2}$. Careful experiments on rare earth salts show good agreement between the calculated and experimental values. Typical numerical values for these rare earth ions range between 1 and 10.

2. At the critical point on the liquid-vapour co-existence curve, we have $(\partial P / \partial V)_T = 0$ (so that κ_T , the compressibility, becomes infinite). We also have $(\partial^2 P / \partial V^2)_T = 0$ (an inflection point on the isotherm). Does this imply that $(\partial^2 V / \partial P^2)_T$ is infinite at the critical point? Give a general answer; check out what happens in the case of the van der Waal's equation(given below).
3. Find the critical parameters T_c , V_c and P_c for the van der Waal's equation of state

$$\left(P + \frac{aN^2}{V^2} \right) (V - Nb) = Nk_B T .$$

Define reduced variables $t \equiv \frac{T - T_c}{T_c}$, $v \equiv \frac{V - V_c}{V_c}$ and $p \equiv \frac{P - P_c}{P_c}$, and express the equation of state in terms of these variables. Simultaneously plot the reduced equation of state, p vs v for three representative values of t ($t > 0$, $t = 0$ and $t < 0$).

Check out this page: <http://sgovindarajan.wikidot.com/notes:phase-transitions>